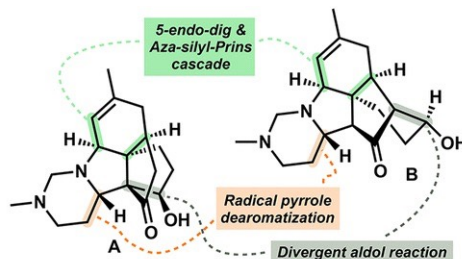


Concise Syntheses of Lycojapomine Alkaloids Enabled by Radical Dearomatization of a Pyrrole

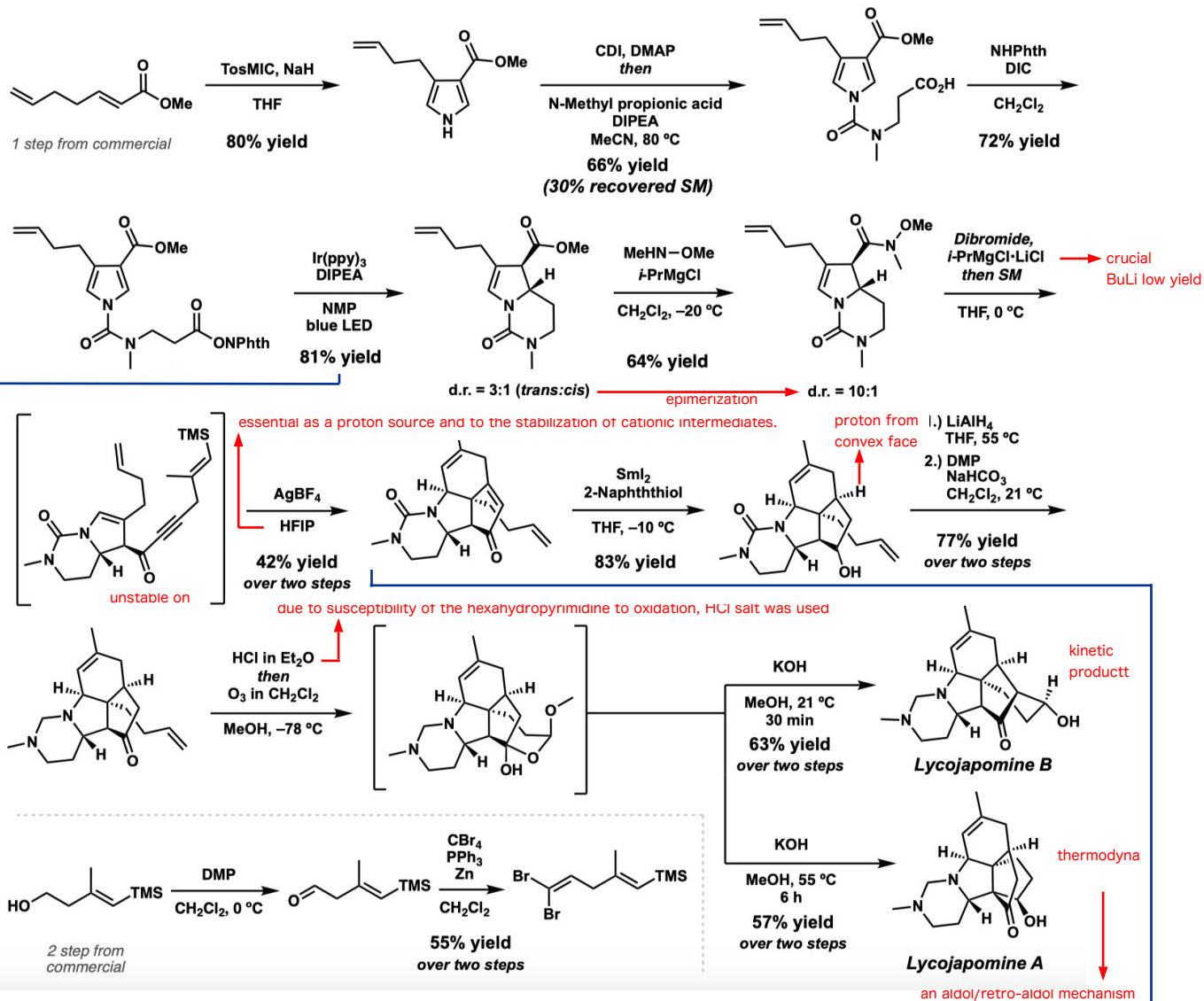
Benjamin M. Gross and Brian M. Stoltz*
J. Am. Chem. Soc. 2025, 147, 20200-20204

a photo-induced radical cyclization to selectively dearomatize a pyrrole

a silver-catalyzed cyclization cascade to annulate two rings and assemble the core carbon framework in a



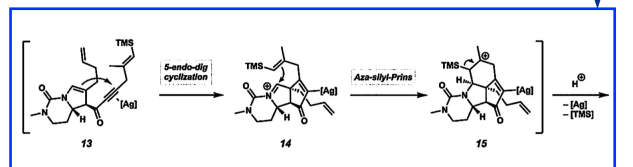
Lycojapomine Alkaloids • First reported total syntheses
• No protecting groups used • 13 steps from commercial starting material



Entry	Catalyst	Additive	Solvent	6 (yield in %)	7 (yield in %)	8 (yield in %)
1 (R=H)	Ir(dFCF ₃ ppy) ₂ (dtbbpy)PF ₆	K ₂ HPO ₄	DMF	-	5	-
2 ^a	Ir(ppy) ₃ • 9 H ₂ O (5 mol %) di(2-picoly)amine (5.1 mol %)	TRIP disulfide (5 mol %), Na ₂ CO ₃ (0.1 equiv)	DCE: H ₂ O (1:1)	-	25	25
3 ^b (R=NPhth)	Eosin Y (10 mol %)	DIPEA (2.5 equiv)	CH ₂ Cl ₂	-	10	-
4 ^c	Ni(acac) ₂ (2.0 equiv)	Zn (50 equiv), LiCl (20 equiv), 50 °C	DMF	-	-	-
5 ^c	Ru(bpy) ₃ (PF ₆) ₂	HE (1.5 equiv), DIPEA (2.0 equiv)	CH ₂ Cl ₂	11	6	15
6 ^c	Ru(bpy) ₃ (PF ₆) ₂	HE (1.5 equiv)	CH ₂ Cl ₂	1	8	22
7 ^c	Ru(bpy) ₃ (PF ₆) ₂	DIPEA (2.0 equiv)	CH ₂ Cl ₂	10	7	11
8 ^c	Ru(bpy) ₃ (PF ₆) ₂	DIPEA (2.0 equiv)	MeCN	8	12	10
9 ^c	Ru(bpy) ₃ (PF ₆) ₂	DIPEA (2.0 equiv)	DMF	-	-	-
10 ^c	Ru(bpy) ₃ (PF ₆) ₂	DIPEA (2.0 equiv)	DMA	11	24	12
11 ^c	Ru(bpy) ₃ (PF ₆) ₂	DIPEA (2.0 equiv)	DMSO	18	32	11
12 ^c	Ru(bpy) ₃ (PF ₆) ₂	DIPEA (2.0 equiv)	NMP	15	23	13
13 ^c	Ru(bpy) ₃ (PF ₆) ₂	Et ₃ N (2.0 equiv)	DMSO	10	17	5
14 ^c	Ru(bpy) ₃ (PF ₆) ₂	Et ₃ N (10.0 equiv)	DMSO	14	24	8
15 ^c	Ir(dibbpy)(ppy) ₂ PF ₆	Et ₃ N (5.0 equiv)	DMSO	25	20	6
16 ^c	Ir(dibbpy)(ppy) ₂ PF ₆	DIPEA (10.0 equiv)	DMSO	42 ^d	35	4
17 ^c	Ir(ppy) ₃	DIPEA (10.0 equiv)	DMSO	53 ^d	33	2
18 ^c	Ir(ppy) ₃	DIPEA (10.0 equiv)	NMP	81	10	3

Reactions were performed with 10 mg of starting material. Yield determined by ¹H-NMR with internal standard. ^aIrradiated with purple LED (390 nm); ^bIrradiated with CFL; ^c1 mol % catalyst used, irradiated with blue LED (450 nm); ^d200 mg scale, isolated yield, d.r. = 3:1.

can be activated at much lower reduction potentials, possibly



Ht: Hantzsch ester

the redox potential of the photocatalyst determined the reaction

Ir(ppy)₃: possesses one of the lower oxidation potentials

UMSU is poorly miscible with DIPEA

Scheme S2: Consideration of pathways for annulation

